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H₂SO₄ • CuSO₄ • CaSiO₃ • H₂S₂O₇ It is the radioactive isotops of Hydrogen: 14. ²₁H • H • H H • It is an alloy of Copper and Tin: 15. Bronze • Brass • Nichrome • German Silver Which pair of the following molecules has the same 16. number of protons? O₂ and N₂ Cl₂ and Br₂ • CH₄ and NH₃ • CO₂ and SO₂ 17. A suspension of droplets of one liquid into another in which it is not soluble is called: smoke • mud • foam • emulsion CHEMISTRY 2018 Time: 2 1/2 Hours Max. Marks: 68 SECTION "B" (SHORT-ANSWER QUESTIONS)(40) NOTE: Answer 10 questions from this section. What is Chemistry? Write names of any six branches of 2. Chemistry. state the Law of Multiple Proportion and explain it with the help of two examples. What is a chemical reaction? Describe decomposition reaction and combustion reaction with at example of each type of reaction. What are "double salts"? Write names of any three such salts with their chemical formulae. Write down four salient features of Mendeleev'd periodic table. How many atoms or molecules are there is? 8g of Sulphur (S) (ii) 8.8g of Carbondioxide (CO₂) (i) (Atomic masses: S = 32, C = 12, O = 16) Reproduce any two of the following chemical reactions in the form of balanced chemical equations: Carbon monoxide reacts with Oxygen gas to form (i) Carbon dioxide

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- Zinc metal reacts with Hydrochloric acid to give Zinc (ii)
- chloride and Hydrogen gas.
- (iii) Potassium Chlorate decomposes on heating into Potassium Chloride and Oxygen gas.
- 9. Define the following: Polar Bond. (iv)
- (i) Isotopes (ii) Electro-negativity (iii) Electron Affinity
- Define pH and Calculate the pH and pOH of 10⁻⁵ molar 10. solution of HNO₃.
- 11. What is meant by Soft water, Hard water & Heavy water? Also give causes of permanent hardness of water.

1+11 كل نثانات: ١٨ <u>صد الف (لازم) (کثیر الانتخابی سوالات)</u> ۱۔ دیۓ ہوئے مکنہ جوابات میں سے ہرایک کیلئے درست جواب کا انتخاب کیجئے۔ (۱۸) (۱) کتاب المتاظر تصنیف ب: * بولی بینا کی * عبدالما لک اسمی کی * الفارانی * این البیشم کی (٢) كسكونا بغير يول والا يوداب-بيب: ٢٠ كمل طفلى ٢٠ جزوى طفيليه ٢٠ مرده خور ٢٠ ٢٠ ماش (٣) دماغ كاده صد جس كالعلق ذبانت، يادداشت وغيره ي ب: (٣) جول كاسب يرونى تحراب: ٢ كيكس ٢ كرولا ٢ ايذروشيم ٢ كاينديم (a) ية كمك حال تبري: A كارنا A ريا C كرائد A الكراك (٢) مالمات كوركت دين ك لخ توانائى استعال ہوتى ب: نفوذ يد واون يد فعال ريل يد فيرفعال ريل 立 يفرون كالعلق ب: * نظام عض * نظام اعصاب * نظام دوران خون * نظام اخراج (4) عصلات كابريون - كمر بخلامد: * ركين * شيندن * ركين اورشيندن * خول (1) لونی پا تچرکی دریافت کرده تحتیک کانام ہے: تابکاریت الم پا<u>ئیرائزیشن</u> اللہ تخیر الم استرلائزیشن . (9) * توارث كى بنيادى اكانى ب: * كروموسوم * ين * يوكيو بازم * مركزه (1.) (۱۱) ماحلیاتی نظام می بزید این:

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CHEMISTRY 2018 **SOLUTION OF NUMERICALS** Q.7. (i) 8 gram of Sulphur (S) Formula: No.of atoms = $\frac{N_A \times given Mass}{N_A \times given Mass}$ Atomic Mass 6.02 x 10²³ x 8 11 32 $\frac{6.02 \times 8}{32} \times 10^{23}$... = 1.505 x 10²³ 11 8.8 gram of Carbon dioxide CO₂ (ii) Formula: N_A x Mass of substance No. of molecules = Formula mass Formula mass of $CO_2 = (1x12) + (2x16) = 44 \text{ gm}$ 6.02 x 10²³ x 8.8 No. of molecules = 6.02 x 8.8 - x 10²³ 44 = 1.204 x 10²³ molecules Q.14 Data: Amount of Current A = 5 Amp. Time t = 1 hour $= 1 \times 60 \times 60 = 3600$ sec. Weight deposited W = ? Z = 0.000329 gm/C SOLUTION: W = ZAT

 in the following ions: (i) O⁻² (ii) Mg⁺² (iii) Cl⁻¹ O¹⁰₈ Mg²⁴₁₂ Cℓ³⁵₁₇ (4) Calculate the number of atoms, present in one gram of Oxygen. (5) Balance the following chemical equations: (i) S + H₂SO₄ → SO₂ + H₂O (ii) C₂H₂ + O₂ → CO₂ + H₂O (iii) NH₃ + Cl₂ → N₂ + HCl (6) Write the names of functional groups and formulae of the following: (i) Ethyl Alcohol (ii) Acetic Acid (iii) Propyl lodide 		$W = 0.000329 \times 5 \times 3600 \text{ sec}$ W = 5.922 gm
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- (8) Define: (i) a.m.u. (ii) mole (iii) mass number
- (9) Write 3 salient features of Mendeleev's periodic table.
- (10) Calculate molarity of solution containing 9.8 g H2So(Ren)

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